```
=> d his

     (FILE 'HOME' ENTERED AT 15:09:29 ON 22 SEP 2006)

     FILE 'REGISTRY' ENTERED AT 15:09:41 ON 22 SEP 2006
L1             0 DISODIUM PRAPERIODATE/CN
L2             0 ?SODIUM  AND PRAPERIODATE
L3             0 H3 I NA2 O6/MF
L4             0 H3INA2O6/MF
L5             0 2/NA AND I AND 3/H AND 6/O
L6             0 SODIUM PARAPERIODATE/CN
L7             1 7790-28-5/RN
L8             1 13940-38-0/RN
L9             1 H5 I O6 . 2 NA /MF
L10            0 H3 I O6 . 2 NA /MF
L11            1 15599-97-0/RN

     FILE 'HCAPLUS' ENTERED AT 15:19:17 ON 22 SEP 2006
L12           37 S L11
L13            0 S L12 AND BLEACH?

     FILE 'REGISTRY' ENTERED AT 15:21:41 ON 22 SEP 2006
L14            1 SODIUM HYPOCHLORITE/CN

     FILE 'HCAPLUS' ENTERED AT 15:21:59 ON 22 SEP 2006
L15        11459 S L14
L16            5 S L12 AND L15

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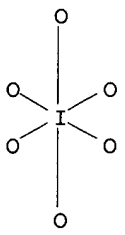
L9 1 H5 I O6 . 2 NA /MF

=> d

L9 ANSWER 1 OF 1 REGISTRY COPYRIGHT 2006 ACS on STN
RN 15599-97-0 REGISTRY
ED Entered STN: 16 Nov 1984
CN Periodic acid (H5IO6), disodium salt (8CI, 9CI) (CA INDEX NAME)
OTHER CA INDEX NAMES:
CN Sodium periodate (Na2H3IO6) (6CI, 7CI)
OTHER NAMES:
CN Sodium paraperiodate (Na2H3IO6)
MF **H5 I O6 . 2 Na**
LC STN Files: CA, CAOLD, CAPLUS, CASREACT, CHEMLIST, GMELIN*, IFICDB, IFIPAT, IFIUDB, TOXCENTER, USPATFULL
(*File contains numerically searchable property data)
Other Sources: EINECS**
(**Enter CHEMLIST File for up-to-date regulatory information)
CRN (10450-60-9)



2 Na

ONE OR MORE TAUTOMERIC DOUBLE BONDS NOT DISPLAYED IN THE STRUCTURE
37 REFERENCES IN FILE CA (1907 TO DATE)
37 REFERENCES IN FILE CAPLUS (1907 TO DATE)
9 REFERENCES IN FILE CAOLD (PRIOR TO 1967)

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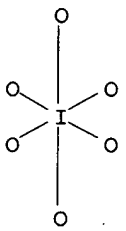
=> d l12 9,17,36 ibib abs hitstr hitind

L12 ANSWER 9 OF 37 HCAPLUS COPYRIGHT 2006 ACS on STN
ACCESSION NUMBER: 1989:127441 HCAPLUS
DOCUMENT NUMBER: 110:127441
TITLE: Sodium hydrogen orthoperiodate Na2H3IO6, a variant of the marcasite structure
AUTHOR(S): Jansen, Martin; Rehr, Anette
CORPORATE SOURCE: Anorg. Chem. Inst., Univ. Bonn, Bonn, D-5300/1, Fed. Rep. Ger.
SOURCE: Zeitschrift fuer Anorganische und Allgemeine Chemie (1988), 567, 95-100
CODEN: ZAACAB; ISSN: 0044-2313
DOCUMENT TYPE: Journal
LANGUAGE: German
AB Single crystals of Na2H3IO6 were grown for the 1st time. X-ray crystal structure detn. (Pnnm; a 469.7(3), b 529.9(2), c 1005.2(6) pm; Z = 2; 296 diffractometer data; Rw = 0.051) shows that I is in an octahedral coordination. Na is surrounded by 6 O atoms in a strongly distorted octahedral arrangement. IO6 and NaO6 groups are linked via common vertex and edges in the sense of the rutile or marcasite type of structure. The corresponding structural relationship is discussed.
IT **15599-97-0P**, Sodium periodate (Na2H3IO6)
RL: SPN (Synthetic preparation); PREP (Preparation)
(prepn. and crystal growth and structure of)
RN 15599-97-0 HCAPLUS
CN Periodic acid (H5IO6), disodium salt (8CI, 9CI) (CA INDEX NAME)



●2 Na

ONE OR MORE TAUTOMERIC DOUBLE BONDS NOT DISPLAYED IN THE STRUCTURE
CC 78-5 (Inorganic Chemicals and Reactions)
IT **15599-97-0P**, Sodium periodate (Na2H3IO6)
RL: SPN (Synthetic preparation); PREP (Preparation)
(prepn. and crystal growth and structure of)

L12 ANSWER 17 OF 37 HCAPLUS COPYRIGHT 2006 ACS on STN
ACCESSION NUMBER: 1979:138401 HCAPLUS
DOCUMENT NUMBER: 90:138401
TITLE: Alkali metal mono- and dibasic periodates
INVENTOR(S): Hillis, James E.; Coker, William P.
PATENT ASSIGNEE(S): Dow Chemical Co., USA
SOURCE: U.S., 3 pp.
CODEN: USXXAM
DOCUMENT TYPE: Patent
LANGUAGE: English
FAMILY ACC. NUM. COUNT: 2
PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE

US 4134967	A	19790116	US 1977-834610	197709 19
AU 7839585	A1	19800313	AU 1978-39585	197809 06
AU 523823	B2	19820819		
CA 1125988	A1	19820622	CA 1978-310714	197809 06
EP 1259	A1	19790404	EP 1978-100885	197809 14
EP 1259	B1	19821013		
R: DE, FR, GB, NL				
JP 54066607	A2	19790529	JP 1978-113701	197809 18
JP 62020985	B4	19870511		
JP 62275002	A2	19871130	JP 1986-276890	198611 21

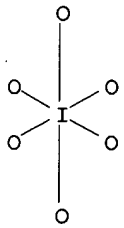
PRIORITY APPLN. INFO.:	US 1977-834609	A	197709 19
	US 1977-834610	A	197709 19

AB The title compds. (useful for converting olefins to the corresponding oxides) were prepd. from tri-, tetra-, and pentabasic alkali metal periodates by treatment with a substance having a relative acidity value (pKa) 3-16 at 0-100° at molar ratios of periodate-acidic substance of 1:1 to 1:1000. Thus, CO2 (20-30 cm3/min) was bubbled 1 h through 2.5 g crude Cs3IO5 in 15 mL water to give 0.9 g CsIO4. Propylene [115-07-1] (15 cm3/min) was passed through a 280-320° glass tube contg. 0.5 g CsIO4 on glass wool to give 28% conversion to propylene oxide [75-56-9] at 312° with selectivity 64%.

IT **15599-97-0P**
RL: IMF (Industrial manufacture); PREP (Preparation)
(manuf. of, from trisodium periodate)

RN 15599-97-0 HCAPLUS

CN Periodic acid (H5IO6), disodium salt (8CI, 9CI) (CA INDEX NAME)

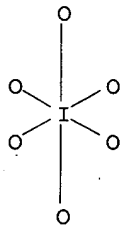


●2 Na

ONE OR MORE TAUTOMERIC DOUBLE BONDS NOT DISPLAYED IN THE STRUCTURE

IC C01B011-22
INCL 423462000
CC 35-2 (Synthetic High Polymers)
Section cross-reference(s): 23, 49
IT **15599-97-0P**
RL: IMF (Industrial manufacture); PREP (Preparation)
(manuf. of, from trisodium periodate)

L12 ANSWER 36 OF 37 HCAPLUS COPYRIGHT 2006 ACS on STN
ACCESSION NUMBER: 1934:52592 HCAPLUS
DOCUMENT NUMBER: 28:52592
ORIGINAL REFERENCE NO.: 28:6382a
TITLE: Periodic acid and periodates. III. Sodium and silver periodates
AUTHOR(S): Partington, James R.; Bahl, Rama K.
SOURCE: Journal of the Chemical Society (1934) 1091-4
CODEN: JCSOA9; ISSN: 0368-1769
DOCUMENT TYPE: Journal
LANGUAGE: Unavailable
AB Na2H3IO6 shows no loss of H2O at 100° in a vacuum, but stronger heating decomposes it according to: 4Na2H3IO6 → 4Na2O + 2I2 + 6H2O + 7O2. The Ag salt behaves as Ag4I2O9.3H2O, losing all of the H2O at 90°.
IT **15599-97-0**, Sodium periodate, Na2H3IO6
(prepn. of)
RN 15599-97-0 HCAPLUS
CN Periodic acid (H5IO6), disodium salt (8CI, 9CI) (CA INDEX NAME)



●2 Na

ONE OR MORE TAUTOMERIC DOUBLE BONDS NOT DISPLAYED IN THE STRUCTURE
CC 6 (Inorganic Chemistry)
IT **15599-97-0**, Sodium periodate, Na2H3IO6
(prepn. of)

=>

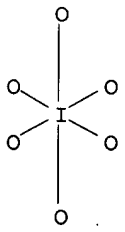
=> d l16 1-6 ibib abs hitstr hitind

L16 ANSWER 1 OF 5 HCAPLUS COPYRIGHT 2006 ACS on STN
ACCESSION NUMBER: 2006:49311 HCAPLUS
DOCUMENT NUMBER: 144:131309
TITLE: Method for manufacture of periodic acid salts having high purity at high yield
INVENTOR(S): Doya, Masaharu; Kurai, Hiroko
PATENT ASSIGNEE(S): Toho Earthtech Inc., Japan
SOURCE: Jpn. Kokai Tokkyo Koho, 8 pp.
CODEN: JKXXAF
DOCUMENT TYPE: Patent
LANGUAGE: Japanese
FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 2006016266	A2	20060119	JP 2004-196891	20040702
PRIORITY APPLN. INFO.:			JP 2004-196891	20040702

AB Manuf. of Na paraperiodate is carried out by oxidn. of I absorption soln., obtained by blowing out process. The thus manufd. Na paraperiodate is further treated with acids for its conversion into Na metaperiodate, followed by its treatment with inorg. K salt for prepn. of potassium metaiodate. Na metaperiodate crystals obtained by the conversion process may be sepd. before addn. of K salt to the mother liquor. Preferable oxidizing agents, acids, etc. are also given.

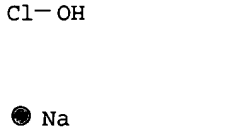
IT **15599-97-0P**
RL: IMF (Industrial manufacture); RCT (Reactant); TEM (Technical or engineered material use); PREP (Preparation); RACT (Reactant or reagent); USES (Uses)
(manuf. of high-purity Na paraperiodate and K metaperiodate therefrom)
RN 15599-97-0 HCAPLUS
CN Periodic acid (H5IO6), disodium salt (8CI, 9CI) (CA INDEX NAME)



●2 Na

ONE OR MORE TAUTOMERIC DOUBLE BONDS NOT DISPLAYED IN THE STRUCTURE
IT **7681-52-9**, Sodium hypochlorite
RL: NUU (Other use, unclassified); USES (Uses)
(oxidizing agent; manuf. of high-purity Na paraperiodate and K metaperiodate therefrom)
RN 7681-52-9 HCAPLUS

CN Hypochlorous acid, sodium salt (8CI, 9CI) (CA INDEX NAME)



CC 49-5 (Industrial Inorganic Chemicals)
IT **15599-97-0P**
RL: IMF (Industrial manufacture); RCT (Reactant); TEM (Technical or engineered material use); PREP (Preparation); RACT (Reactant or reagent); USES (Uses)
(manuf. of high-purity Na paraperiodate and K metaperiodate therefrom)
IT **7681-52-9**, Sodium hypochlorite 7722-84-1, Hydrogen peroxide, uses 7782-50-5, Chlorine, uses
RL: NUU (Other use, unclassified); USES (Uses)
(oxidizing agent; manuf. of high-purity Na paraperiodate and K metaperiodate therefrom)

L16 ANSWER 2 OF 5 HCAPLUS COPYRIGHT 2006 ACS on STN
ACCESSION NUMBER: 2004:799519 HCAPLUS
DOCUMENT NUMBER: 141:298144
TITLE: Method for preparing disodium para-periodate
INVENTOR(S): Yoshikawa, Kouji
PATENT ASSIGNEE(S): Sumitomo Chemical Company, Limited, Japan
SOURCE: PCT Int. Appl., 12 pp.
CODEN: PIXXD2
DOCUMENT TYPE: Patent
LANGUAGE: Japanese
FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
WO 2004083117	A1	20040930	WO 2004-JP3491	200403 16
W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW				
RW: BW, GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IT, LU, MC, NL, PL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG				
JP 2004300023	A2	20041028	JP 2004-74111	200403 16
EP 1619167	A1	20060125	EP 2004-720972	200403 16
R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO, MK, CY, AL, TR, BG, CZ, EE, HU, PL, SK				
CN 1761614	A	20060419	CN 2004-80007113	200403

PRIORITY APPLN. INFO.: JP 2003-75248 A 16
200303
19

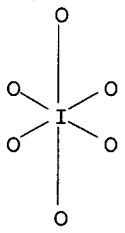
WO 2004-JP3491 W
200403
16

AB The method includes reacting mixt. of NaIO3, HIO3, and NaClO with NaOH (1-3 mol for total IO3-) at pH 5-10. The obtained Na2H3IO6 is contacted with acid at pH 2-2.5 to produce NaIO4. Na2H3IO6 is produced by the safe and simple method without Cl2 gas.

IT **15599-97-0P**, Sodium paraperiodate (Na2H3IO6)
RL: PUR (Purification or recovery); PREP (Preparation)
(method for prepg. disodium para-periodate)

RN 15599-97-0 HCAPLUS

CN Periodic acid (H5IO6), disodium salt (8CI, 9CI) (CA INDEX NAME)



●2 Na

ONE OR MORE TAUTOMERIC DOUBLE BONDS NOT DISPLAYED IN THE STRUCTURE

IT **7681-52-9**, Sodium hypochlorite
RL: RCT (Reactant); RACT (Reactant or reagent)
(method for prepg. disodium para-periodate)

RN 7681-52-9 HCAPLUS

CN Hypochlorous acid, sodium salt (8CI, 9CI) (CA INDEX NAME)

Cl—OH

● Na

IC ICM C01B011-22

CC 49-5 (Industrial Inorganic Chemicals)

IT 7790-28-5P, Sodium metaperiodate **15599-97-0P**, Sodium paraperiodate (Na2H3IO6)
RL: PUR (Purification or recovery); PREP (Preparation)
(method for prepg. disodium para-periodate)

IT **7681-52-9**, Sodium hypochlorite 7681-55-2, Sodium iodate 7782-68-5, Iodic acid
RL: RCT (Reactant); RACT (Reactant or reagent)
(method for prepg. disodium para-periodate)

REFERENCE COUNT: 5 THERE ARE 5 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L16 ANSWER 3 OF 5 HCAPLUS COPYRIGHT 2006 ACS on STN

ACCESSION NUMBER: 1987:166050 HCAPLUS
DOCUMENT NUMBER: 106:166050
TITLE: Silver halide photographic processing wastewater treatment kit containing halite/hypohalite/perhalate
INVENTOR(S): Kuze, Satoru; Koboshi, Shigeharu; Matsushima, Yoko
PATENT ASSIGNEE(S): Konishiroku Photo Industry Co., Ltd., Japan
SOURCE: Jpn. Kokai Tokkyo Koho, 17 pp.
CODEN: JKXXAF
DOCUMENT TYPE: Patent
LANGUAGE: Japanese
FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 61240238	A2	19861025	JP 1985-83276	19850417
JP 05054930	B4	19930813		

PRIORITY APPLN. INFO.: JP 1985-83276 19850417

AB A wastewater treatment kit for Ag halide photog. processing soln. in an automatic developing system that has no feed/wastewater piping and uses no rinsing water, is characterized by comprising a preadjusted prepn. of ≥1 compds. selected from perhalates, halites, and hypohalites to render the COD of the effluent substantially 0.

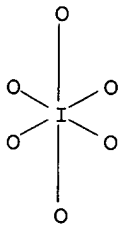
IT **7681-52-9**, Sodium hypochlorite **15599-97-0**, Sodium periodate
RL: USES (Uses)
(wastewater treatment kit contg., for silver halide photog. processing soln. in automatic developing system)

RN 7681-52-9 HCAPLUS
CN Hypochlorous acid, sodium salt (8CI, 9CI) (CA INDEX NAME)

Cl—OH

● Na

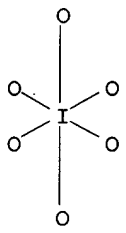
RN 15599-97-0 HCAPLUS
CN Periodic acid (H5IO6), disodium salt (8CI, 9CI) (CA INDEX NAME)



●2 Na

ONE OR MORE TAUTOMERIC DOUBLE BONDS NOT DISPLAYED IN THE STRUCTURE
IC ICM G03C005-00
CC 74-2 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)
Section cross-reference(s): 60
IT **7681-52-9**, Sodium hypochlorite 7758-19-2, Sodium chlorite
15599-97-0, Sodium periodate
RL: USES (Uses)
(wastewater treatment kit contg., for silver halide photog. processing soln. in automatic developing system)

L16 ANSWER 4 OF 5 HCAPLUS COPYRIGHT 2006 ACS on STN
ACCESSION NUMBER: 1961:1253 HCAPLUS
DOCUMENT NUMBER: 55:1253
ORIGINAL REFERENCE NO.: 55:204b-c
TITLE: Complex manganese periodates
AUTHOR(S): Lister, M. W.; Yoshino, Y.
CORPORATE SOURCE: Univ. Toronto
SOURCE: Canadian Journal of Chemistry (1960), 38, 1291-9
CODEN: CJCHAG; ISSN: 0008-4042
DOCUMENT TYPE: Journal
LANGUAGE: Unavailable
AB Reaction of $Na_2H_3IO_6$ in dil. HNO_3 with $MnCl_2$ and a basic soln. of NaClO yielded $Na_7H_4Mn(IO_6)_3 \cdot 17H_2O$. When KIO_4 and KClO were used, $K_7H_4Mn(IO_6)_3 \cdot 8H_2O$ was formed. Iodometric and potentiometric titrations, redn. with SO_2, and magnetic susceptibilities all indicate that the compds. contain quadrivalent Mn. Solns. of the Na salt spontaneously decomp. to yield $NaMnO_4$. The mechanism of decompn. is not clear.
IT **15599-97-0**, Sodium periodate, $Na_2H_3IO_6$
(reaction with $MnCl_2$ and NaClO)
RN 15599-97-0 HCAPLUS
CN Periodic acid (H5IO6), disodium salt (8CI, 9CI) (CA INDEX NAME)



●2 Na

ONE OR MORE TAUTOMERIC DOUBLE BONDS NOT DISPLAYED IN THE STRUCTURE
IT **7681-52-9**, Sodium hypochlorite
(reaction with $MnCl_2$ and $Na_2H_3IO_6$)
RN 7681-52-9 HCAPLUS
CN Hypochlorous acid, sodium salt (8CI, 9CI) (CA INDEX NAME)

Cl—OH

● Na

CC 6 (Inorganic Chemistry)

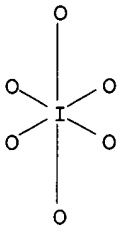
IT **15599-97-0**, Sodium periodate, Na2H3IO6
(reaction with MnCl2 and NaClO)
IT **7681-52-9**, Sodium hypochlorite
(reaction with MnCl2 and Na2H3IO6)

L16 ANSWER 5 OF 5 HCAPLUS COPYRIGHT 2006 ACS on STN
ACCESSION NUMBER: 1955:38999 HCAPLUS
DOCUMENT NUMBER: 49:38999
ORIGINAL REFERENCE NO.: 49:7445i,7446a-c
TITLE: Paper partition chromatography of halogen salts
AUTHOR(S): Servigne, Yvette
SOURCE: Compt. rend. (1954), 239, 272-4
DOCUMENT TYPE: Journal
LANGUAGE: Unavailable
AB Mixts. of chlorates, bromates, and iodates of the alkali metals were chromatographed onto paper from iso-PrOH (75% by vol.) the partition taking approx. 16 hrs. The chromatograms were developed by steaming the strips of paper first in an all-glass app. contg. diphenylamine in concd. H3PO4 to bring out the blue-green stains of the bromates and iodates and, second, in another app. contg. diphenylamine in concd. HCl to bring out the indigo-blue stains of the chlorates. The chlorate was found at the top of the chromatogram, followed in order by the bromate and the iodate. It was found that 3.5 γ of one of the salts could be detected in as much as 91 γ of the other two. Rf = 0.68 for KClO3, 0.47-0.149 for KBrO3, and 0.20-0.21 for KIO3. The iodate in soln. alone showed 2 adjacent stains, corresponding to Rf = 0.28 and 0.21, which merge into one when the iodate is in soln. with the other salts; the stain corresponding to Rf = 0.28 is presumably accounted for by the presence of small amts. of periodates. Supporting evidence for this presumption is given by the exhibition of Rf values for mixts. of periodates and iodates identical to those of the iodates when present in soln. alone; the periodate probably decomp. considerably in contact with the cellulose of the paper. The paraperiodate, Na2H3IO6, showed only one stain corresponding to Rf = 0.28. Mixts. of Na chlorite, hypochlorite, and chloride were also examd.
IT **7681-52-9**, Sodium hypochlorite
(chromatography of)
RN 7681-52-9 HCAPLUS
CN Hypochlorous acid, sodium salt (8CI, 9CI) (CA INDEX NAME)

Cl—OH

● Na

IT **15599-97-0**, Sodium periodate, Na2H3IO6
(detection of)
RN 15599-97-0 HCAPLUS
CN Periodic acid (H5IO6), disodium salt (8CI, 9CI) (CA INDEX NAME)



●2 Na

ONE OR MORE TAUTOMERIC DOUBLE BONDS NOT DISPLAYED IN THE STRUCTURE
CC 7 (Analytical Chemistry)
IT 7647-14-5, Sodium chloride **7681-52-9**, Sodium hypochlorite
7758-19-2, Sodium chlorite
(chromatography of)
IT 3811-04-9, Potassium chlorate 7758-01-2, Potassium bromate
7758-05-6, Potassium iodate **15599-97-0**, Sodium periodate, Na2H3IO6
(detection of)

=>

=> d l12 ti 1-37

L12 ANSWER 1 OF 37 HCAPLUS COPYRIGHT 2006 ACS on STN
TI Method for manufacture of periodic acid salts having high purity at high yield

L12 ANSWER 2 OF 37 HCAPLUS COPYRIGHT 2006 ACS on STN
TI Method for preparing disodium para-periodate

L12 ANSWER 3 OF 37 HCAPLUS COPYRIGHT 2006 ACS on STN
TI Preparation of oxalide via 9-oxononanoic acids

L12 ANSWER 4 OF 37 HCAPLUS COPYRIGHT 2006 ACS on STN
TI Process for production of 3,3-dimethyl-2-formylcyclopropanecarboxylic acid derivatives for the preparation of pyrethroids

L12 ANSWER 5 OF 37 HCAPLUS COPYRIGHT 2006 ACS on STN
TI Slime inhibitors containing periodic acid and slime prevention

L12 ANSWER 6 OF 37 HCAPLUS COPYRIGHT 2006 ACS on STN
TI Manufacture of sodium metaperiodate

L12 ANSWER 7 OF 37 HCAPLUS COPYRIGHT 2006 ACS on STN
TI Manufacture of periodic acid alkali metal salts

L12 ANSWER 8 OF 37 HCAPLUS COPYRIGHT 2006 ACS on STN
TI Manufacture of disodium trihydrogenparaperiodate

L12 ANSWER 9 OF 37 HCAPLUS COPYRIGHT 2006 ACS on STN
TI Sodium hydrogen orthoperiodate Na2H3IO6, a variant of the marcasite structure

L12 ANSWER 10 OF 37 HCAPLUS COPYRIGHT 2006 ACS on STN
TI Silver halide photographic processing wastewater treatment kit containing halite/hypohalite/perhalate

L12 ANSWER 11 OF 37 HCAPLUS COPYRIGHT 2006 ACS on STN
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TI Chemical influence on the decay constant of iodine-125

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TI Chemical effect of the iodine-125 decay constant

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TI Thermal stability of analytical standards. VII

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TI Thermal stability of analytical standards. VI

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TI Dielectric behavior of periodates

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